ON THE EVIDENCE FOR INTRAMOLECULAR LIGAND EXCHANGE IN PENTACOORDINATE CARBONATE(IV) ANION: IMPLICATIONS FOR THE MECHANISM OF

BIMOLECULAR NUCLEOPHILIC SUBSTITUTION AT SATURATED CARBON.

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(Received in USA 27 August 1975; received in UK for publication 6 October 1975) The spectacular finding¹ that <u>cis</u>-3-ethoxycyclobutyl brosylate (<u>1</u>) undergoes S_N^2 displacement by iodide and bromide ion with retention of configuration, as predicted² on theoretical grounds, understandably attracted the attention of the chemical community,³ for this observation served to cast an entirely new light on the classical S_N^2 mechanism: according to the experimental¹ and theoretical² evidence, the reaction proceeds through an intermediate pentacoordinate carbonate(IV) anion of trigonal bipyramidal geometry, which is capable of undergoing intramolecular ligand exchange, <u>i.e.</u> Berry pseudorotation or turnstile rotation.

We now report that halide displacement reactions on <u>cis</u>- and <u>trans-1</u> in fact proceed cleanly with inversion of configuration.

<u>Cis-1</u>, ⁴ prepared from <u>cis</u>-3-ethoxycyclobutanol (<u>2</u>), ⁵ was converted in quantitative yield to <u>trans</u>-3-ethoxycyclobutyl acetate by treatment with potassium acetate in anhydrous dimethyl sulfoxide containing dicyclohexyl-18-crown-6.

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Reaction of the acetate with methylmagnesium iodide gave <u>trans-2</u>, which was converted to <u>trans-1</u>.⁴ Reaction of <u>cis-</u> and <u>trans-1</u> with lithium bromide under the previously described conditions¹ yielded 3-ethoxycyclobutyl bromides (<u>3</u>),⁶ which were analyzed by glc on a 10% Carbowax 20M column, using mesitylene as an internal standard. The results are graphically displayed in Figure 1.

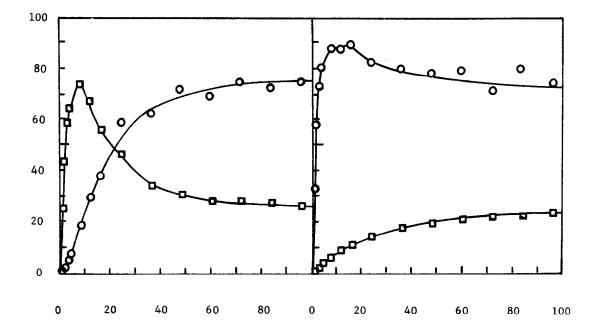


Figure 1. Percent conversion of 3-ethoxycyclobutyl brosylates (<u>1</u>) into bromides (<u>3</u>) (ordinate) as a function of reaction time, in hours (abscissa), starting with <u>cis-1</u> (left) and <u>trans-1</u> (right). Circles: <u>cis-3</u>. Squares: <u>trans-3</u>.

As seen by inspection of the Figure, initially stereospecific conversion of <u>cis-1</u> into <u>trans-3</u> and of <u>trans-1</u> into <u>cis-3</u> is followed by epimerization⁹ to an equilibrium mixture of <u>cis-</u> and <u>trans-3</u> in a <u>ca</u>. 3:1 ratio.¹⁰ We hasten to emphasize that these stereochemical results, far from surprising, were in fact fully anticipated since it has long been known that S_N^2 displacement reactions on related cyclobutane ring systems proceed with inversion of configuration,^{5,11} and that cyclobutyl halides are epimerized by halide ion.^{7a}

The present work obviously invalidates all evidence for the claim¹ that there exist carbon centered S_N^2 reaction intermediates with lifetimes long enough for intramolecular ligand exchange. Furthermore, since retention was predicted² for S_N^2 displacements at cyclobutyl carbon, the failure of that prediction necessarily vitiates the theoretical model² on which the prediction was based. In particular, it is now appropriate to question the claim,² based on CND0/2 calculations, that electronegative substituents prefer the equatorial position in the putative trigonal bipyramidal carbon centered S_N^2 intermediates. Further comments are reserved for our detailed paper on the subject.

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References and Notes

- T. ElGomati, D. Lenoir, and I. Ugi, <u>Angew. Chem.</u>, <u>87</u>, 66 (1975); <u>Angew</u>. <u>Chem. internat. Ed.</u>, <u>14</u>, 59 (1975).
- P.D. Gillespie and I. Ugi, <u>Angew. Chem.</u>, <u>83</u>, 493 (1971); <u>Angew. Chem.</u> <u>internat. Ed.</u>, <u>10</u>, 503 (1971).
- 3. <u>Nachr. Chem. Techn.</u>, <u>22</u>, 409 (1974); <u>Chem. and Ind.</u>, 272 (1975).
- 4. I. Lillien and L. Handloser, Tetrahedron Lett., 1213 (1970).
- 5. J.B. Sieja, <u>J. Amer. Chem. Soc.</u>, <u>93</u>, 130 (1971).
- 6. The configurational identities of <u>cis</u>- and <u>trans-1</u>, <u>2</u>, and <u>3</u> were determined by nmr, according to established precedent.^{4,5,7} The configurational assignments⁴ for <u>1</u> and <u>2</u> were confirmed by X-ray analysis on <u>cis-1</u>.⁸
- 7. (a) K.B. Wiberg and G.M. Lampman, <u>J. Amer. Chem. Soc.</u>, <u>88</u>, 4429 (1966).
 (b) I. Lillien and R.A. Doughty, <u>ibid.</u>, <u>89</u>, 155 (1967).
- 8. J.F. Blount, unpublished work.
- In an independent experiment, reaction of <u>trans-3</u> with lithium bromide in refluxing acetone led to the same equilibrium mixture.
- 10. Similar results were obtained in reactions of $\underline{1}$ with sodium iodide in acetone.
- K.B. Wiberg and G.L. Nelson, <u>Tetrahedron Lett</u>., 4385 (1969);
 P.v.R. Schleyer, P. LePerchec, and D.J. Raber, <u>ibid</u>., 4389 (1969).